# Configuration of the active Mg-ATP complex in protein kinase C reaction

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To probe the active site structure of protein kinase C stereochemical studies were carried out by using ATP $\beta$ S. The enzyme utilizes either one of the diastereomers ( $S_P$  and  $R_P$ ) of ATP $\beta$ S almost equally well as a substrate. This result contrasts with that for cyclic AMP-dependent protein kinase, suggesting that the topography of the nucleotide-binding site is significantly different between the two kinases.

Phosphorylation Protein kinase C ATPBS Stereospecificity

#### 1. INTRODUCTION

Phosphorylation and dephosphorylation of proteins are one of the major mechanisms by which intracellular events are regulated [1]. Over 30 protein kinase isozymes are now known and they phosphorylate target proteins to modulate their physiological activities. Despite such biological importance, little is known about the mechanism by which these enzymes mediate the phosphoryl transfer to an acceptor substrate. Bolen and collaborators [2] reported that cAMP-dependent protein kinase (protein kinase A) exhibits high stereospecificity for the \( \Delta \) configuration of Mg-ATP as substrate. An intriguing question is raised as to whether other protein kinases show the same stereospecificity. Here, we have investigated the stereochemical preference of Ca2+-dependent, phospholipid-activated protein kinase (protein kinase C) [3] for the diastereomers of adenosine 5'-O-(2-thiotriphosphate) (ATP $\beta$ S), phosphorothioate analogue of ATP. It was found that, unlike protein kinase A, this enzyme utilizes both diastereomers almost equally well as substrate, suggesting a considerable difference in the topography of the nucleotide-binding site of both kinases.

## 2. MATERIALS AND METHODS

## 2.1. Materials

Chemicals and gels used in this work were obtained from the following sources: histone type IIIS, phosphatidylserine and diolein, Sigma; DEAE-cellulose and P81 phosphocellulose paper, Whatman; phenyl-Sepharose and AH-Sepharose, Pharmacia;  $[\gamma^{-32}P]ATP$ , Amersham.  $[\gamma^{-32}P]ATP\beta S_P$  and  $R_P$  isomers were prepared as in [4] and [5], respectively. Protein kinase C was purified from rat brain as in [6]. Homogeneous protein kinase A catalytic subunit was prepared from bovine heart as detailed in [7,8].

## 2.2. Protein kinase assay

Protein kinase C was assayed at 30°C essentially as in [6]. The assay mixture contained the following components: 20 mM Tris-HCl (pH 7.5), 1 mM 2-mercaptoethanol, 5 mM magnesium acetate, 200  $\mu$ g/ml histone IIIS, 0.1 mM calcium chloride, 10  $\mu$ g/ml phosphatidylserine, 1  $\mu$ g/ml diolein, 50  $\mu$ M [ $\gamma$ -<sup>32</sup>P]ATP or ATP $\beta$ S. The reaction was started by the addition of enzyme and terminated by placing an aliquot on a piece of P81 filter paper [9].

# 3. RESULTS

It was found that a phosphorothioate analogue of ATP, ATP $\beta$ S, serves as a phosphoryl donor substrate for protein kinase C. The relative effectiveness of the  $S_P$  and  $R_P$  diastereomers was compared under standard assay conditions with histone H1 as the phosphoryl acceptor substrate. As shown in table 1, both isomers exhibited almost identical reactivity, which is about 15% that of ATP. This result is in sharp contrast to that for protein kinase A, which adopts the  $S_P$  isomer as the preferred substrate by a large margin [2]. To rule out the possibility that the data on protein kinase C are due to an artifact of our ATP $\beta$ S preparations, stereochemical studies were also carried out for protein kinase A. When compared in terms of  $V_{\text{max}}/K_{\text{m}}$ , the  $R_{\text{P}}$  isomer was only 1.3% as active as the  $S_P$  isomer. Protein kinase C possesses no activity with Cd<sup>2+</sup> as the activating metal. This behavior is also different from that of protein kinase A [2].

To gain insight into the origin of the above stereochemical (in) discrimination a detailed kinetic analysis was attempted. However, initial rate data for ATP $\beta$ S did not follow the Henri-Michaelis-Menten equation closely under the various conditions tested, where similar data for ATP can be analyzed to give a  $K_{\rm m}$  of 6.3  $\pm$  0.1  $\mu$ M [10,11]. Nonetheless, the  $K_{\rm m}$  for either diastereomer of ATP $\beta$ S appears to lie in the range 1–10  $\mu$ M and is of similar magnitude. This suggests that both stereoisomers behave almost identically not only in binding but also in catalytic processes in the protein kinase C reaction.

Table 1
Stereochemical preference of protein kinase C for the diastereomers of ATPβS

Nucleotide	Velocity <sup>a</sup> (nmol/min per mg)	Relative rate (%)
ATP	$67.5 \pm 6.1 (n = 3)$	100
ATP $\beta$ S $S_P$	$10.5 \pm 0.7 \ (n = 5)$	16
ATPβS R <sub>P</sub>	$8.6 \pm 1.7 (n = 5)$	13

<sup>&</sup>lt;sup>a</sup> For assay conditions see section 2°

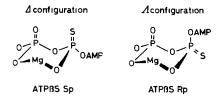


Fig. 1. Configurations of the  $\beta, \gamma$ -bidentate Mg-ATP $\beta$ S complexes.

### 4. DISCUSSION

Our results reveal that protein kinase C utilizes either diastereomer of ATP $\beta$ S equally well as a substrate. Since the magnesium complex of the  $S_P$  isomer of ATP $\beta$ S assumes a  $\Delta$  configuration, while the corresponding complex of the  $R_P$  isomer takes on a  $\Lambda$  configuration (fig.1) [12], these data may imply that the enzyme accepts either geometry of the  $\beta$ , $\gamma$ -bidentate Mg-ATP complex to its active site. This result is in contrast to that for protein kinase A, which prefers predominantly the  $\Delta$  configuration as the active substrate [13].

Although the exact  $K_{\rm m}$  values could not be determined for ATP $\beta$ S, it was suggested that ATP $\beta$ S  $S_{\rm P}$  and  $R_{\rm P}$  isomers and ATP possess a nearly identical  $K_{\rm m}$ . This indicates that the difference in the overall reactivity of ATP and its analogues (table 1) reflects mainly the catalytic process. Sulfur substitution on the  $\beta$ -P of ATP could diminish the interaction of nucleotide substrate with an active site residue(s), leading to a decreased reactivity relative to ATP.

In summary, the present data suggest that the spatial disposition of the nucleotide-binding site of protein kinases A and C may be considerably different. In line with this notion is the observation that the two enzymes behave distinctly toward isoquinolinesulfonamides which are supposed to bind to the nucleotide-binding site [14]. It would be interesting to see whether protein kinase C retains a sequence like -Gly-X-Gly-X-X-Gly- which is presumed to constitute an ATP-binding site of many proteins including the protein kinase family [15].

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## REFERENCES

- Krebs, E.G. and Beavo, J.A. (1979) Annu. Rev. Biochem. 48, 923-959.
- [2] Bolen, D.W., Stingelin, J., Bramson, H.N. and Kaiser, E.T. (1980) Biochemistry 10, 1176-1182.
- [3] Nishizuka, Y. (1984) Nature 308, 693-697.
- [4] Stingelin, J., Bolen, D.W. and Kaiser, E.T. (1980)J. Biol. Chem. 255, 2022-2025.
- [5] Romanisk, P.J. and Eckstein, F. (1981) J. Biol. Chem. 256, 7322-7328.
- [6] Kitano, T., Go, M., Kikkawa, U. and Nishizuka, Y. (1986) Methods Enzymol., in press.
- [7] Demaille, J.G., Peters, K.A. and Fischer, E.H. (1977) Biochemistry 16, 3080-3086.

- [8] Armstrong, R.N., Kondo, H., Granot, J., Kaiser, E.T. and Mildvan, A.S. (1979) Biochemistry 18, 1230-1238.
- [9] Glass, D.B., Masaracchia, R.A., Feranisco, J.R. and Kemp, B.E. (1978) Anal. Biochem. 87, 566-575.
- [10] Wise, B.C., Glass, D.B., Chou, C.-H.J., Raynor, R.L., Katoh, N., Schatzman, R.C., Turner, R.S., Kibler, R.F. and Kuo, J.F. (1982) J. Biol. Chem. 257, 8489-8495.
- [11] Kikkawa, U., Takai, Y., Minakuchi, R., Inohara, S. and Nishizuka, Y. (1982) J. Biol. Chem. 257, 13341-13348.
- [12] Jaffe, E. and Cohn, M. (1978) J. Biol. Chem. 253, 4823-4825.
- [13] Granot, J., Mildvan, A.S., Brown, E., Kondo, H., Bramson, H.N. and Kaiser, E.T. (1979) FEBS Lett. 103, 265-269.
- [14] Hidaka, H., Inagaki, M., Kawamoto, S. and Sasaki, Y. (1984) Biochemistry 23, 5036-5041.
- [15] Takio, K., Blumenthal, D.K., Edelman, A.M., Walsh, K.A., Krebs, E.G. and Titani, K. (1985) Biochemistry 24, 6028-6037.